

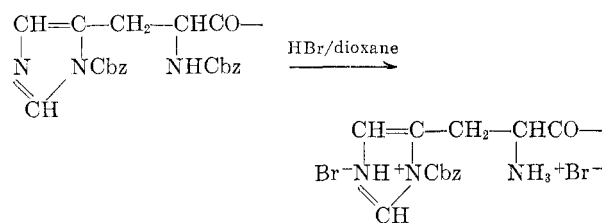
New Method for Synthesis of Peptides Containing Histidine

Sir:

In recent years it has been suggested that a histidine residue occupies a site of the biological activity in some naturally occurring peptides and proteins. Synthesis of histidine peptides is, therefore, very important for the study of protein biochemistry. However, the synthetic procedure has not been substantially improved for lack of the simple and satisfactory method to protect selectively the imidazole function of histidine.¹

The purpose of this communication is to describe a new method for the synthesis of peptides containing histidine by selective blocking of the imidazole ring with a carbobenzoxy group.

Akabori *et al.*^{2,3} have reported that *N*(α),*N*(Im)-dicarbobenzoxyhistidine is an excellent starting material in the synthesis of histidyl peptides. We have now found that the carbobenzoxy group linked at the imidazole-nitrogen (*N*(Im)-cbz) of histidine is surprisingly resistant to the treatment with hydrogen bromide in dioxane or in glacial acetic acid.



N(α),*N*(Im)-Di-cbz-histidine methyl ester hydrochloride (II) (m.p. 121–122.5° dec.⁴),⁴ which was derived from di-cbz-histidine (I)⁵ with thionyl chloride in methanol, and *N*(α),*N*(Im)-di-cbz-histidylphenylalanine methyl ester (III) (m.p. 136–137°) were treated with 35–40% (w/w)-hydrogen bromide in dioxane at room temperature for 30 min. to give *N*(Im)-cbz-histidine methyl ester dihydrobromide (IV) [94% yield, m.p. 167–167.5° dec., $\lambda_{\text{max}}^{\text{CH}_3\text{OH}}$ 234 m μ (ϵ 3140)]. *Anal.* Calcd. for C₁₅H₁₉N₃O₄Br₂: C, 38.75; H, 4.12; N,

9.04; Br, 34.4. Found: C, 37.23; H, 4.44; N, 9.43; Br, 35.03] and *N*(Im)-cbz-histidylphenylalanine methyl ester dihydrobromide (V) [99% yield, m.p. 126.5–127.5° dec., $\lambda_{\text{max}}^{\text{CH}_3\text{OH}}$ 234 m μ (ϵ 3300)]. *Anal.* Calcd. for C₂₄H₂₈N₄O₅Br₂: C, 47.15; H, 4.61; N, 9.15; Br, 26.15. Found: C, 46.87; H, 4.75; N, 9.05; Br, 26.26], respectively. These *N*(α)-free derivatives are reactive to ninhydrin and their absorptions at 234 m μ were extremely depressed by adding sodium methoxide within a few minutes. This phenomenon shows the presence of an *N*(Im)-cbz group.⁶ On the other hand, these derivatives can react somewhat slowly with the Pauly reagent different from the disubstituted histidine analogs. This fact may show that the *N*(Im)-cbz group is very unstable in an alkaline medium in the absence of an *N*(α)-substituting function.

IV or V could, however, be safely neutralized with dilute aqueous ammonia in methylene chloride at 0°. The resulting free esters were coupled with cbz-glycine by means of the carbodiimide method⁷ to yield cbz-glycyl-*N*(Im)-cbz-histidine methyl ester (VI) [85% yield, m.p. 74–75.5°, $[\alpha]_{\text{D}}^{29} +24.5^\circ$ (ethyl acetate), $\lambda_{\text{max}}^{\text{CH}_3\text{OH}}$ 236 m μ (ϵ 3755)]. *Anal.* Calcd. for C₂₅H₂₆N₄O₇: C, 60.7; H, 5.31; N, 11.3. Found: C, 60.61; H, 5.45; N, 11.43] and cbz-glycyl-*N*(Im)-cbz-histidylphenylalanine methyl ester (VII) [90% yield, m.p. 154–154.5°, $[\alpha]_{\text{D}}^{18} -8.6^\circ$ (methanol), $\lambda_{\text{max}}^{\text{CH}_3\text{OH}}$ 236 m μ (ϵ 3655)]. *Anal.* Calcd. for C₃₄H₃₅N₅O₈: C, 63.7; H, 5.51; N, 10.9. Found: C, 63.69; H, 5.63; N, 10.84]. In a similar fashion we have obtained cbz-glycyl-*N*(Im)-cbz-histidylleucine methyl ester (VIII) [82% yield, m.p. 152–152.5°, $[\alpha]_{\text{D}}^{18} -11.6^\circ$ (methanol), $\lambda_{\text{max}}^{\text{CH}_3\text{OH}}$ 236 m μ (ϵ 3610)]. *Anal.* Calcd. for C₃₁H₃₇N₅O₈: C, 61.3; H, 6.14; N, 11.5. Found: C, 61.39; H, 6.31; N, 11.44], *N*(α), *N*(Im)-dicbz-histidyl-*N*(Im)-cbz-histidylphenylalanine benzyl ester (IX) [81% yield, m.p. 138–140°, $[\alpha]_{\text{D}}^{22} -17.9^\circ$ (methanol), $\lambda_{\text{max}}^{\text{CH}_3\text{OH}}$ 236 m μ (ϵ 7830)]. *Anal.* Calcd. for C₅₂H₄₉N₇O₁₀: C, 67.1; H, 5.79; N, 10.5. Found: C, 67.05; H, 5.41; N, 10.69] and formyl- γ -methylglutamyl-*N*(Im)-cbz-histidylphenylalanine benzyl ester (X) [87% yield, m.p. 165–165.5° dec., $[\alpha]_{\text{D}}^{24} -12.1^\circ$ (dimethylformamide), $\lambda_{\text{max}}^{\text{CH}_3\text{OH}}$ 236 m μ (ϵ 3415)]. *Anal.* Calcd. for C₃₇H₃₉N₅O₉: C, 63.7; H, 5.63; N, 10.02. Found: C, 63.79; H, 5.90; N, 10.36] which has a sequence contained in corticotropin and MSH. These histidine peptide derivatives (VI to X) are negative to both ninhydrin and the Pauly reagent and have moderate solubilities in common organic solvents.

The treatment of VII with hydrogen bromide/dioxane or with 2 equivalent amounts of sodium hydroxide gave glycyl-*N*(Im)-cbz-histidylphenyl-

(1) Benzyl [D. Theodoropoulos, *J. Org. Chem.*, **21**, 1550 (1956)] or trityl group [G. C. Stelakatos, D. M. Theodoropoulos, and L. Zervas, *J. Am. Chem. Soc.*, **81**, 2884 (1959)] has been used for the protection of the imidazole function but there are some defects for practical purposes.

(2) S. Akabori, K. Okawa, and F. Sakiyama, *Nature*, **181**, 772 (1958).

(3) F. Sakiyama, K. Okawa, T. Yamakawa, and S. Akabori, *Bull. Chem. Soc. Japan*, **31**, 926 (1958).

(4) Amino acids used are of L-configuration and all melting points are uncorrected.

(5) This could be crystallized in granules from a concentrated ethyl acetate solution [m.p. 90.5–92° dec., $[\alpha]_{\text{D}}^{19} +29.1^\circ$ (ethyl acetate)]. *Anal.* Calcd. for C₂₂H₂₁N₃O₅: N, 9.93. Found: N, 9.86]. This crystal is readily soluble in methanol and from the solution is soon separated a second form of crystal containing one molecule of methanol, m.p. 105–107° dec.^{3, 6}

(6) A. Patchornik, A. Berger, and E. Katchalski, *J. Am. Chem. Soc.*, **79**, 6416 (1957).

(7) J. C. Sheehan and G. P. Hess, *J. Am. Chem. Soc.*, **77**, 1067 (1955).

alanine methyl ester dihydrobromide (XI) [$\lambda_{\max}^{\text{CH}_3\text{OH}}$ 234 $m\mu$ (ϵ 3640). *Anal.* Calcd. for $\text{C}_{26}\text{H}_{31}\text{N}_5\text{O}_6\text{Br}_2$: C, 46.7; H, 4.68; N, 10.01; Br, 23.9. Found: C, 45.73; H, 4.99; N, 10.40; Br, 22.59] or cbz-glycyl-histidylphenylalanine (XII) [77.5% yield, m.p. 200° dec., $[\alpha]_{\text{D}}^{30} + 15.4^\circ$ (methanol). *Anal.* Calcd. for $\text{C}_{25}\text{H}_{27}\text{N}_5\text{O}_6$: C, 60.8; H, 5.52; N, 14.2. Found: C, 60.51; H, 5.65; N, 14.11], respectively. The *N*(Im)-cbz group may be removed also with catalytic hydrogenation.

From these results it has been demonstrated that *N*(Im)-cbz-histidine derivatives, which can be prepared by the simple and convenient procedure, are of potential utility as intermediates in the synthesis of histidine peptides.

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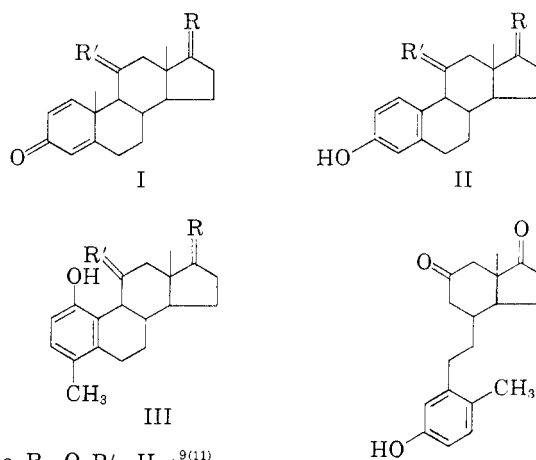
An Aromatization Reaction of A Cross-Conjugated Dienone System with Zinc

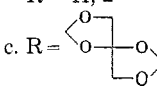
Sir:

We wish to report a new A-ring aromatization reaction of the dienone system with zinc under mild conditions. When refluxed with zinc in pyridine¹ or ethylene glycol, androstan-1,4,9(11)-triene-3,17-dione (Ia) undergoes A-ring aromatization with elimination of the angular methyl group to form Δ^9 -estrone (IIa)² in excellent yield (75%). In a similar manner the treatment of 17 α ,21-dihydroxypregnan-1,4,9(11)-triene-3,20-dione acetate (Ib; 21-acetate) or its BMD derivative (Ic) in pyridine yielded, respectively, A-ring aromatic corticoids,³ IIb 21-acetate, (yield: 35%), m.p. 210–212°, $[\alpha]_{\text{D}}^{20} + 174^\circ$ (dioxane), $\lambda_{\max}^{\text{CH}_3\text{OH}}$ 263, 298 $m\mu$ (ϵ 18,000, 3,100), $\lambda_{\max}^{\text{Nujol}}$ 814 cm^{-1} . (*Anal.* Calcd. for $\text{C}_{22}\text{H}_{26}\text{O}_5$: C, 71.33; H, 7.08. Found: C, 71.48; H, 6.98); IIb, m.p. 248–250°, $[\alpha]_{\text{D}}^{20} + 176^\circ$ (dioxane); IIb, 3,21-diacetate, m.p. 188–190°, $[\alpha]_{\text{D}}^{20} + 136^\circ$ (chloroform), or IIc (yield: 72%), m.p. 247–248°, $[\alpha]_{\text{D}}^{20} + 31^\circ$ (dioxane), which was converted to IIb by acetic acid hydrolysis. 3-Keto-1,4,6-triene (Id)

gave Δ^6 -estrone (IIId)⁴ (yield, 10–15%) in the same reaction.

In the case of 3-keto-1,4-dienes or their C-11 substituted derivatives different rearrangement products were obtained. Treatment of Ie with zinc in pyridine provided a mixture of *p*-cresol type rearrangement product (IIIe)⁵ (yield, 80%) and estrone (IIe) (yield, 4%). The 11 β -hydroxy compound (If) yielded IIIf, m.p. 223–224°, $[\alpha]_{\text{D}}^{15} + 249^\circ$ (chloroform), $\lambda_{\max}^{\text{CH}_3\text{OH}}$ 282–286 $m\mu$ (ϵ 2,340), methyl ether, m.p. 213–214°, $[\alpha]_{\text{D}}^{15} + 303^\circ$ (chloroform), which was confirmed by conversion to the IIIe methyl ether by dehydration and hydrogenation. However, the 11-keto compound (Ig) suffered rupture of C₉-C₁₀ bond with concomitant A-ring aromatization to give the 9//10 seco compound (IV), m.p. 212–214°. It was identical in all respects with an authentic specimen of IV obtained from Ig by pyrolysis.⁶



- a. R = O, R' = H, $\Delta^{9(11)}$
 b. R = α -OH, $-\text{CO}-\text{CH}_2\text{OH}$,
 R' = H, $\Delta^{9(11)}$
 c. R =  R' = H, $\Delta^{9(11)}$
 d. R = O, R' = -H, -H, Δ^6
 e. R = O, R' = -H, -H
 f. R = O, R' = β -OH, -H
 g. R = R' = O

Santonin was converted to desmethyl-desmotroposantonin (V) (yield, 40%), m.p. 223–227°, $[\alpha]_{\text{D}}^{20} + 115^\circ$ (chloroform), $\lambda_{\max}^{\text{CH}_3\text{OH}}$ 286.5 $m\mu$ (ϵ 2,820), $\lambda_{\max}^{\text{Nujol}}$ 811 cm^{-1} , NMR⁷ τ , 7.60 ppm. (one benzenoid methyl) (*Anal.* Calcd. for $\text{C}_{14}\text{H}_{16}\text{O}_3$: C, 72.39; H, 6.94. Found: C, 72.10; H, 6.93), acetate, m.p. 144–146° (*Anal.* Calcd. for $\text{C}_{16}\text{H}_{18}\text{O}_4$: C, 70.05; H, 6.61. Found: C, 70.08; H, 6.50.), by the same reaction in pyridine. The structure of V was confirmed by the palladium-charcoal dehydrogenation which led to 1-methyl-7-ethyl-

(4) St. Kaufman, J. Pataki, G. Rosenkranz, J. Romo, and C. D. Djerassi, *J. Am. Chem. Soc.*, **72**, 4531, 4534 (1950).

(5) A. S. Dreiding and A. Voltman, *J. Am. Chem. Soc.*, **76**, 537 (1954).

(6) B. J. Magerlein and J. A. Hogg, *Tetrahedron*, **2**, 80 (1958).

(7) τ values were calculated assuming τ chloroform (solvent) = 2.75 ppm.

(1) Pyridine containing ca. 10 mole equivalents of water was used.

(2) B. J. Magerlein and J. A. Hogg, *J. Am. Chem. Soc.*, **80**, 2220 (1958).

(3) B. J. Magerlein and J. A. Hogg, *J. Am. Chem. Soc.*, **80**, 2226 (1958).